

# Phase transformations in LiH2PO4 (LDP) revealed by Raman spectroscopy

Rawdha Dekhili, T.H. Kauffmann, Hassen Aroui, Marc D. Fontana

### ► To cite this version:

Rawdha Dekhili, T.H. Kauffmann, Hassen Aroui, Marc D. Fontana. Phase transformations in LiH2PO4 (LDP) revealed by Raman spectroscopy. Solid State Communications, 2018, 279, pp.22-26. 10.1016/j.ssc.2018.04.016 . hal-01803857

## HAL Id: hal-01803857 https://centralesupelec.hal.science/hal-01803857

Submitted on 31 May 2018  $\,$ 

**HAL** is a multi-disciplinary open access archive for the deposit and dissemination of scientific research documents, whether they are published or not. The documents may come from teaching and research institutions in France or abroad, or from public or private research centers. L'archive ouverte pluridisciplinaire **HAL**, est destinée au dépôt et à la diffusion de documents scientifiques de niveau recherche, publiés ou non, émanant des établissements d'enseignement et de recherche français ou étrangers, des laboratoires publics ou privés.

# Accepted Manuscript

Phase transformations in LiH<sub>2</sub>PO<sub>4</sub> (LDP) revealed by Raman spectroscopy

Rawdha Dekhili, Thomas H. Kauffmann, Hassen Aroui, Marc D. Fontana

PII: S0038-1098(18)30227-8

DOI: 10.1016/j.ssc.2018.04.016

Reference: SSC 13432

To appear in: Solid State Communications

Received Date: 22 December 2017

Revised Date: 23 April 2018

Accepted Date: 29 April 2018

Please cite this article as: R. Dekhili, T.H. Kauffmann, H. Aroui, M.D. Fontana, Phase transformations in LiH<sub>2</sub>PO<sub>4</sub> (LDP) revealed by Raman spectroscopy, *Solid State Communications* (2018), doi: 10.1016/j.ssc.2018.04.016.

This is a PDF file of an unedited manuscript that has been accepted for publication. As a service to our customers we are providing this early version of the manuscript. The manuscript will undergo copyediting, typesetting, and review of the resulting proof before it is published in its final form. Please note that during the production process errors may be discovered which could affect the content, and all legal disclaimers that apply to the journal pertain.



### Phase transformations in LiH<sub>2</sub>PO<sub>4</sub> (LDP) revealed by Raman spectroscopy

Rawdha Dekhili<sup>a,b,c,d</sup>, Thomas H. Kauffmann<sup>a,b</sup>, Hassen Aroui<sup>c,d</sup> and Marc D. Fontana<sup>a,b</sup>

<sup>a</sup>Université de Lorraine, Laboratoire Matériaux Optiques, Photonique et Systèmes, EA 4423 - 2, rue Edouard Belin, 57070 Metz, France.

<sup>b</sup>CentraleSupélec, Laboratoire Matériaux Optiques, Photonique et Systèmes, EA 4423 - 2, rue Edouard Belin, 57070 Metz, France.

<sup>c</sup>Université de Tunis Elmanar, Faculté des Sciences de Tunis, 2092 El Manar Tunis, Tunisia.

<sup>d</sup>Université de Tunis, Laboratoire Dynamique Moléculaire et Matériaux Photoniques, Ecole Nationale Supérieure d'Ingénieurs de Tunis – 5 av. Taha Hussein, 1008 Tunis, Tunisia.

#### Abstract

We investigate Raman scattering measurements on polycrystalline  $LiH_2PO_4$  (LDP) between 30 and 240°C, in order to detect phase transformations. The thermal behavior of main lines in different wavenumber ranges is discussed and analyzed. Raman spectra show two anomalies at 176°C and 210°C, reflecting changes in chemical bonds, and are attributed to progressive polymerization, in agreement with earlier electrical measurements.

Keywords : Raman spectroscopy, vibrational modes, phosphate compounds, phase transformations

#### **1. Introduction**

The family of phosphate compounds with the general formula  $MX_2PO_4$  (M = K, Rb, NH<sub>4</sub>, Cs, Tl, Li, X = H, D) is the object of many investigations thanks to several physical and chemical properties interesting for basic research and practical applications [1,2]. Among these compounds, the most widely studied is potassium dihydrogen phosphate (KDP) for ferroelectric, piezoelectric, and electro-optic properties [3,4].

Much less investigated,  $LiH_2PO_4$  (LDP) has a particular interest for its relatively high protonic conductivity [5-7]. The structure of LDP consists into tetrahedral groups PO<sub>4</sub> and LiO<sub>4</sub> which are bonded together by oxygen ions. Lee et al [8] have recorded Raman data on LDP between 70 and 300 K but did not observe any change in spectra at low temperature. They did not investigate high temperature range but by means of electrical conductivity and dielectric constant measurements they found two anomalies at 178°C (Tp<sub>1</sub>) and 196°C (Tp<sub>2</sub>) [9]. They interpreted them as due to super-protonic phase transformations induced by polymerization at Tp<sub>1</sub> and Tp<sub>2</sub>. Therefore we have performed Raman spectroscopy measurements at high temperature with a special attention to the temperature range 170–220°C.

#### 2. Experimental

LDP polycrystals were prepared by employing a slow evaporation method using  $Li_2CO_3$  (purity of 99.99%), phosphoric acid (purity of 95%) and distilled water as starting materials according to the reaction

 $Li_2CO_3(solid) + 2 H_3PO_4(liquid) \iff 2 Li(H_2PO_4)_2(solid) + CO_2(gas) + 2 H_2O(liquid)$ 

The resulting solution was stirred at room temperature. A quantity of the obtained homogeneous solution was kept in Petri dishes at ambient temperature during a few days; the final product consists into transparent cm-size samples.

Raman spectra were recorded on a polycrystalline sample using confocal Raman microscope LabRAM HR Evolution (*Horiba*) with an exciting laser line of 532 nm. The system leads to a

spectral resolution of 1 cm<sup>-1</sup> using a 1800 grooves per millimeter grating and is equipped with an ultra-low frequency module to detect wavenumbers down to 5 cm<sup>-1</sup>. The laser light was focused through a 50x objective (*Olympus*) at a distance above 1 cm from the sample placed into a temperature stage (*Linkam*). Raman spectra of LDP were recorded in the spectral range of 5–3500 cm<sup>-1</sup> as a function of temperature from 30°C up to 240°C.

#### 3. Results and analysis



Figure 1. Raman spectrum of  $LiH_2PO_4$  recorded at room temperature in the region from 5 to 1400 cm<sup>-1</sup>. Arrows indicate the lines which are particularly analyzed.

The Raman spectrum recorded at room temperature is shown in figure 1 and is found to be relatively similar to spectra reported earlier [8,10]. We distinguish different frequency ranges: below 300 cm<sup>-1</sup> are lying the bands generally associated to external modes; between 300 and 1100 cm<sup>-1</sup> the lines generally attributed to LiO<sub>4</sub> and PO<sub>4</sub> internal vibrations and above 1100 cm<sup>-1</sup> very weak structures corresponding to OH vibrations.

We report and discuss the temperature dependence of Raman spectra especially between 170 and 220°C. For this we describe and analyze the temperature behavior of the most intense lines lying in 4 different wavenumber ranges. These peaks are indicated by arrows in figure 1.



Figure 2. Raman spectra of  $LiH_2PO_4$  recorded at different temperatures in the region from 50 to 95 cm<sup>-1</sup>. In the insert is reported the temperature dependence of the position of the line lying at 85 cm<sup>-1</sup> at room temperature.

In figures 2 and 3 are shown the Raman spectra recorded at different temperatures, in the ranges corresponding to external modes and high-frequency internal modes respectively.

We note a continuous change of the profile when heating between 30 and 170°C, with a clear down-shift of the maxima positions, and a decrease of intensity of all considered lines. The diminishing of the maximum position with increasing temperature is expected regarding the thermal dilatation. The chemical bonds are expected to increase on heating, so that the force constant diminishes and subsequently the frequency decreases. Above 170°C the maximum position of main lines displays two breakdowns in its monotonous decrease on heating (see insert of figure 2).

Likewise, regarding the intensity we remark that for larger temperature, all lines (see figures 2 and 3) exhibit a large discontinuity in their behavior, with an abrupt decrease of intensity, between 210 and 220°C. In particular, we can point out the large diminishing of the line at about 85 cm<sup>-1</sup> which is very intense at room temperature.



Figure 3. Raman spectra of  $LiH_2PO_4$  recorded at four different temperatures in the region from 860 to 960 cm<sup>-1</sup>.

To attempt to rely more precisely the Raman data with the phase transformations, we report in figure 4 the temperature dependence of the maximum intensity of the main lines.



Figure 4. Temperature dependence of the maximum intensity of four different Raman peaks of  $LiH_2PO_4$  (the lines are labeled according to their position at room temperature).

#### ACCEPTED MANUSCRIPT

The intensity of all peaks shows firstly a monotonous decrease on heating, but the line at 85 cm<sup>-1</sup> which is nearly constant, and then two clear breakdowns in their dependence. These two clear anomalies appear around 176°C and 210°C, close – but slightly shifted – to Tp<sub>1</sub> and Tp<sub>2</sub> reported by Lee by means of electrical measurements [9].

It is to be noted that, on the one hand the external modes, and on the other hand the internal modes sensitive to P-O and Li-O bonds, are all affected by a structural rearrangement. These features reflect a change of the whole structure of the material, and constitute therefore an evidence of phase transformations.

An additional remarkable feature is given in figure 5 by the temperature dependence of the quasielastic scattering, i.e. the signal beside the Rayleigh and external mode bands (see the insert). In this figure is reported the integrated intensity between 5 and 30 cm<sup>-1</sup> as function of temperature. This plot shows a large peak around 176°C and a second small maximum around 210°C. The spectral range below 20 cm<sup>-1</sup> covers (even partly) the collective protonic and relaxational motions of PO<sub>4</sub> and LiO<sub>4</sub>.



*Figure 5. Temperature dependence of the integrated intensity between 5 and 30 cm<sup>-1</sup> of the quasielastic scattering reported in the insert.* 

The quasi-elastic scattering can be related to the dielectric response in the teraHertz or subteraHertz frequency range. Thus, as the Stokes Raman intensity is proportional to the imaginary part of the dielectric susceptibility  $\chi''(\omega)$  [11], the integration of the Raman intensity over the frequency range 0–30 cm<sup>-1</sup> provides the real part of the susceptibility  $\chi'(\omega)$  at 1 THz. This last feature can be related to dielectric permittivity measurements of Lee et al [9] and thus shows some consistencies with them. This demonstrates that the whole Raman spectrum is found to be affected by structural transformations.

As shown in figure 6 the thermal change affects also the very high wavenumber range, corresponding to the OH bands. At first we note a shift on heating of the whole spectrum, which is

mainly caused by the increase of the background (see level around 1400 cm<sup>-1</sup>, i.e. far from any Raman band). This can be related to the loss of transparency of the sample which becomes milky above 200°C. The most remarkable change occurs at 240°C. The narrow lines around 1200 cm<sup>-1</sup>, and the broad stretching bands between 2600 and 3600 cm<sup>-1</sup> disappear, and are replaced by a very intense and wide signal. This huge profile change occurs simultaneously with the appearance of water on the surface of the sample. In fact the sample is notably damaged by heating process. This is confirmed by the spectrum obtained after cooling which completely differs from this previously recorded on heating (see figure 6). Therefore the transformations invoked above are clearly irreversible. A thermal decomposition does exist at high temperature with water releasing. The progressive let of water is consistent with the chemical reactions proposed by Lee et al. [9]



Figure 6. High wavenumber Raman spectra of  $LiH_2PO_4$  recorded during a heating-cooling process. Spectra at 240°C (heating) and 210°C (cooling) are vertically shifted for clearness.

The thermal behavior of the largest peak around 888 cm<sup>-1</sup> (position at room temperature) reported in figure 4 corroborates this analysis. This line which shows the strongest decrease of intensity on heating, corresponds to the internal vibration  $v_1$ (P-OH). This bond is therefore affected by the continuous releasing of water, explaining more abrupt changes at Tp<sub>1</sub> and Tp<sub>2</sub>, suggested to be the onset of dimerization and oligomerization respectively.

#### 4. Conclusion

On heating up to 170°C, Raman lines show an expected decrease of their maximum position. At higher temperature their intensity shows a breakdown in their monotonous behavior with temperature. Two anomalies were detected in the temperature dependence around 176°C and 210°C, for all main Raman lines. Consequently these features reflect the large modifications of the chemical bonds at high temperature, and therefore transformations of the whole structure. These

results are consistent with electrical data reported by Lee et al [9] and corroborate their interpretation.

#### References

[1] Yaroslav Shchur. Lattice dynamics study and specific heat of  $CsH_2PO_4$  and  $CsD_2PO_4$  », Physical Review B 74 (2006): 054301. doi.org/10.1103/PhysRevB.74.054301

[2] Fabrício Mendes Souza. Electrical Conductivity in the KDP, ADP, and  $K_{1-x}(NH_4)_xH_2PO_4$  Crystals », Materials Research 20, (avril 2017): 532-37, doi.org/10.1590/1980-5373-mr-2016-0603.

[3] Leimin Deng, Jun Duan. Xiaoyan Zeng et al. A study on dual laser beam separation technology of KDP crystal », International Journal of Machine Tools and Manufacture. 72, Supplement C (2013): 1-10. doi.org/10.1016/j.ijmachtools.2013.05.001.

[4] S. Goma, C. M. Padma, C. K. Mahadevan. Dielectric parameters of KDP single crystals added with urea, Materials Letters 60, no 29 (2006): 3701-5. doi.org/10.1016/j.matlet.2006.03.092.

[5] Kwang-Sei Lee, In-Hwan Oh, Jin Jung Kweon et al. Crystal Growth and morphology of LiH<sub>2</sub>PO<sub>4</sub>. Materials Chemistry and physics 136 (2012): 802-808. doi.org/10.1016/j.matchemphys.2012.08.001.

[6] Jin Jung Kweon, Kyu Won Lee, Cheol Eui Lee, et al. Impedance spectroscopy of the superprotonic conduction in LiH<sub>2</sub>PO<sub>4</sub>, Applied Physics Letters 101, 1 (2012):101, 012905, doi.org/10.1063/1.4732790.

[7] Jin Jung Kweon, Kyu WonLee, Kwang-SeiLee et al. Rotating-frame nuclear magnetic resonance study of the superprotonic conduction in LiH<sub>2</sub>PO<sub>4</sub>, Solid State Communications. 2013 (171). 5-7. doi.org/10.1016/j.ssc.2013.07.022
[8] Kwang-Sei Lee, Jae-Hyeon Ko, Joonhee Moon et al. Raman spectroscopic study of LiH<sub>2</sub>PO<sub>4</sub>, Solid State Communications 145, 9 (2008): 487-92. doi.org/10.1016/j.ssc.2007.12.011.

[9] Kwang-Sei Lee, Joonhee Moon, Jisu Lee et al. High-temperature phase transformations in LiH2PO4 and possible solid-state polymerization. Solid State Communications. 147 (2008) 74-77. doi.org/http://dx.doi.org/10.1016/j.ssc.2008.04.011.

[10] T.Rhimi, N.Nguili, S.Guermazi et al. Structural and vibrational study a new potassium lithium dihydrogenphosphate  $KLi(H_2PO_4)_2$ , Vibrational Spectroscopy 86 (2016): 50-60, doi.org/10.1016/j.vibspec.2016.05.008. [11] Y. Tominaga and H. Urabe. Central component of y(xy)x Raman spectra in paraelectric  $KH_2PO_4$ . Solid State Communications, 41, 7, (1982) 561-564, 0038-1098/82/070561.

- Use of Raman spectroscopy to detect phase transformations
- Phase transformations in a new material (LDP) among KDP family compounds
- Link between Raman data and dielectric measurements